

Atomic Ions

H	
hydrogen	H ⁺
hydride	H ⁻
hydronium	H ₃ O ⁺
hydroxide	OH ⁻

CN	
cyanide	CN ⁻
cyanate	OCN ⁻
thiocyanate	SCN ⁻

B		C		N		O		F	
borate	BO ₃ ³⁻	carbide	C ₂ ²⁻ or C ⁴⁻	nitride	N ³⁻	oxide	O ²⁻	fluoride	F ⁻
		carbonate	CO ₃ ²⁻	azide	N ₃ ⁻	peroxide	O ₂ ²⁻		
		bicarbonate	HCO ₃ ⁻	nitrite	NO ₂ ⁻				
		oxalate	C ₂ O ₄ ²⁻	nitrate	NO ₃ ⁻				
		acetate	C ₂ H ₃ O ₂ ⁻	amide	NH ₂ ⁻				
				ammonium	NH ₄ ⁺				
Al		Si		P		S		Cl	
aluminum	Al ³⁺	silicate	SiO ₄ ⁴⁻	phosphide	P ³⁻	sulfide	S ²⁻	chloride	Cl ⁻
				phosphite	PO ₃ ³⁻	sulfite	SO ₃ ²⁻	hypochlorite	ClO ⁻
				phosphate	PO ₄ ³⁻	sulfate	SO ₄ ²⁻	chlorite	ClO ₂ ⁻
				hydrogen phosphate	HPO ₄ ²⁻	bisulfite	HSO ₃ ⁻	chlorate	ClO ₃ ⁻
				dihydrogen phosphate	H ₂ PO ₄ ⁻	bisulfate	HSO ₄ ⁻	perchlorate	ClO ₄ ⁻
						thiosulfate	S ₂ O ₃ ²⁻		
Cr		Mn		As				Br	
chromate	CrO ₄ ²⁻	permanganate	MnO ₄ ⁻	Arsenide	As ³⁻			bromide	Br ⁻
dichromate	Cr ₂ O ₇ ²⁻			Arsenite	AsO ₃ ³⁻			hypobromite	BrO ⁻
				Arsenate	AsO ₄ ³⁻			bromite	BrO ₂ ⁻
								bromate	BrO ₃ ⁻
								perbromate	BrO ₄ ⁻
								I	
								iodide	I ⁻
								triiodide	I ₃ ⁻
								hypoiodite	IO ⁻
								iodite	IO ₂ ⁻
								iodate	IO ₃ ⁻
								periodate	IO ₄ ⁻